





- Ensure all glassware is free of moisture and oxygen. Dry glassware in an oven, transfer it to a desiccator to cool, then assemble glassware in the fume hood and purge with inert gas.
- Transfer pyrophoric liquids following manufacturer's recommendations, the [Aldrich Technical Information Bulletin No AL-134](#), the technique outlined by Schwindeman et al. (2002), or your laboratory-specific procedure.
- Transfer pyrophoric solids (e.g., finely-divided metals) in a dry, inert atmosphere in a glovebox.
- Dispersions of pyrophoric solid in mineral oil (e.g., potassium hydride or sodium hydride) can be weighed as follows:<sup>2</sup>
  - Weigh out required amount of dispersion and seal in a flask under nitrogen.
  - Add dry hexanes via syringe, swirl and let solids settle.
  - Slowly syringe off hexanes.
  - Carefully discard into a separate flask containing isopropanol.
  - Repeat rinse



- After rinsing, leave the container open in the back of a fume hood for at least one week before disposal. Deface the label.
- Never leave a container with residue of pyrophoric material open to the atmosphere.
- All materials (e.g., disposable gloves, bench paper) that contain or are contaminated with pyrophoric materials must be decontaminated and disposed as hazardous waste.
- Submit a request through the online system for hazardous waste pickup.

## Emergency procedures

Ensure proper labeling of hazardous waste. See EazEn4.3 (0e)-00.8 (f Tmik)(e)-39 0 Td19 ( h)( w)-e0e 1 T8 18 1f0.004 Tc Td19 ( h)



## References

1. Schwindeman, J. A., C. J. Woltermann, and R. J. Letchford. (2002). Safe handling of organolithium compounds in the laboratory. *Journal of Chemical Health & Safety* 9(3):6.
2. UCLA Chemistry & Biochemistry, Procedures for Safe Use of Pyrophoric solids (2/2009) [https://www.chemistry.ucla.edu/sites/default/files/safety/sop/SOP\\_Pyrophoric.pdf](https://www.chemistry.ucla.edu/sites/default/files/safety/sop/SOP_Pyrophoric.pdf) (accessed June 6 2019).
3. AL-134 Handling air sensitive reagents. (1997) [https://www.sigmaaldrich.com/content/dam/sigma-aldrich/docs/Aldrich/Bulletin/al\\_techbull\\_al134.pdf](https://www.sigmaaldrich.com/content/dam/sigma-aldrich/docs/Aldrich/Bulletin/al_techbull_al134.pdf) (accessed June 6 2019).
4. National Research Council (U.S.). (2011). *Prudent practices in the laboratory: Handling and management of chemical hazards*. Washington, D.C: National Academies Press.
5. Bowen, J. E. (2019, September 3). *Metal fires require knowledge of proper extinguishing agents*. Fire Engineering. <https://www.fireengineering.com/leadership/metal-fires-require-knowledge-of-proper-extinguishing-agents/>.