reaction coordinate [4,8–11]. The committor allows calculation of the reaction rate from a one-dimensional description [12] and identifies the transition-state ensemble as states making up the  $q_{ab} = 0.5$  isocommittor surface [13].

making up the  $q_{\phi}=0.5$  isocommittor surface [13]. In this Letter, we derive a novel information-theoretic justification of the committor as the reaction coordinate.

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selected based on the trajectory outcome  $S_+$  [the next mesostate (A or B) visited by the system] and origin  $S_-$  (the mesostate most recently visited by the system). This partitions the supertrajectory into four trajectory subensembles, each with particular  $s \equiv (S_-$ 

where  $p(s) = \sum_{\phi} p(\phi; s)$  is the marginal probability

Combining Eqs. (1), (8a), and (9b) gives the full TPE entropy production as a function of the forward committor,

 $p_R \Sigma$ 

for which one has physical intuition can provide a lowdimensional model that allows increased insight into the reaction mechanism.

More concretely, this connection we have established between transition-path theory and stochastic thermodynamics suggests a novel method for rigorously grounded inference of reaction coordinates: generate an ensemble of transition paths using transition-path sampling [6,31] or related algorithms [32–34]; estimate entropy production along chosen coordinates [35–37] or identify linear combinations of coordinates producing the most entropy using dissipative components analysis [38]; use these most dissipative coordinates to enhance sampling of transition paths, and through further iteration identify system coordinates producing the most entropy in the transition-path ensemble and hence of most relevance to the reaction.

Machine-learning approaches to solve for high-dimensional committor coordinates