

## **Electrocatalytic CO<sub>2</sub> Reduction to Formate for Renewable Energy Storage Applications**

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Electroreduction of CO<sub>2</sub> to formate (HCOO<sup>-</sup>) is one of the most intensely studied pathways for CO<sub>2</sub> utilization. The product formate salt (e.g., Na or K) solution can be utilized as a convenient energy vector for applications in fuel cells and batteries. The bifunctional (or bidirectional) electrode for reversible CO<sub>2</sub> conversion to formate is essential for development of novel energy storage technologies such as CO<sub>2</sub> redox flow batteries and reversible formate fuel cells. The presentation reviews recent advances in electrocatalysis and electrode engineering for CO<sub>2</sub> reduction to formate and formate electro-oxidation to carbonate. A major advantage of CO<sub>2</sub> based electrochemical energy storage is the several orders of magnitude lower cost of CO<sub>2</sub> compared to any metal (e.g., lithium, cobalt, nickel, zinc) used in batteries. Furthermore, it offers a new and viable business case for CO<sub>2</sub> utilization, namely, long-duration and large-scale energy storage with a market that is estimated to grow fifteen-fold between 2022 and 2030. Results will be discussed showing the evolution of the CO<sub>2</sub> redox flow battery from fundamental electrocatalysis to the engineering of 100 cm<sup>2</sup> electrode area flow cells. The experimental results are contextualized within a techno-economic analysis assessing the viability of different electrochemical CO<sub>2</sub> utilization pathways.